



## Americas hydrogen and syngas technical training seminar

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Industrial SMR Aftertreatment Solutions  
Leo DeRita

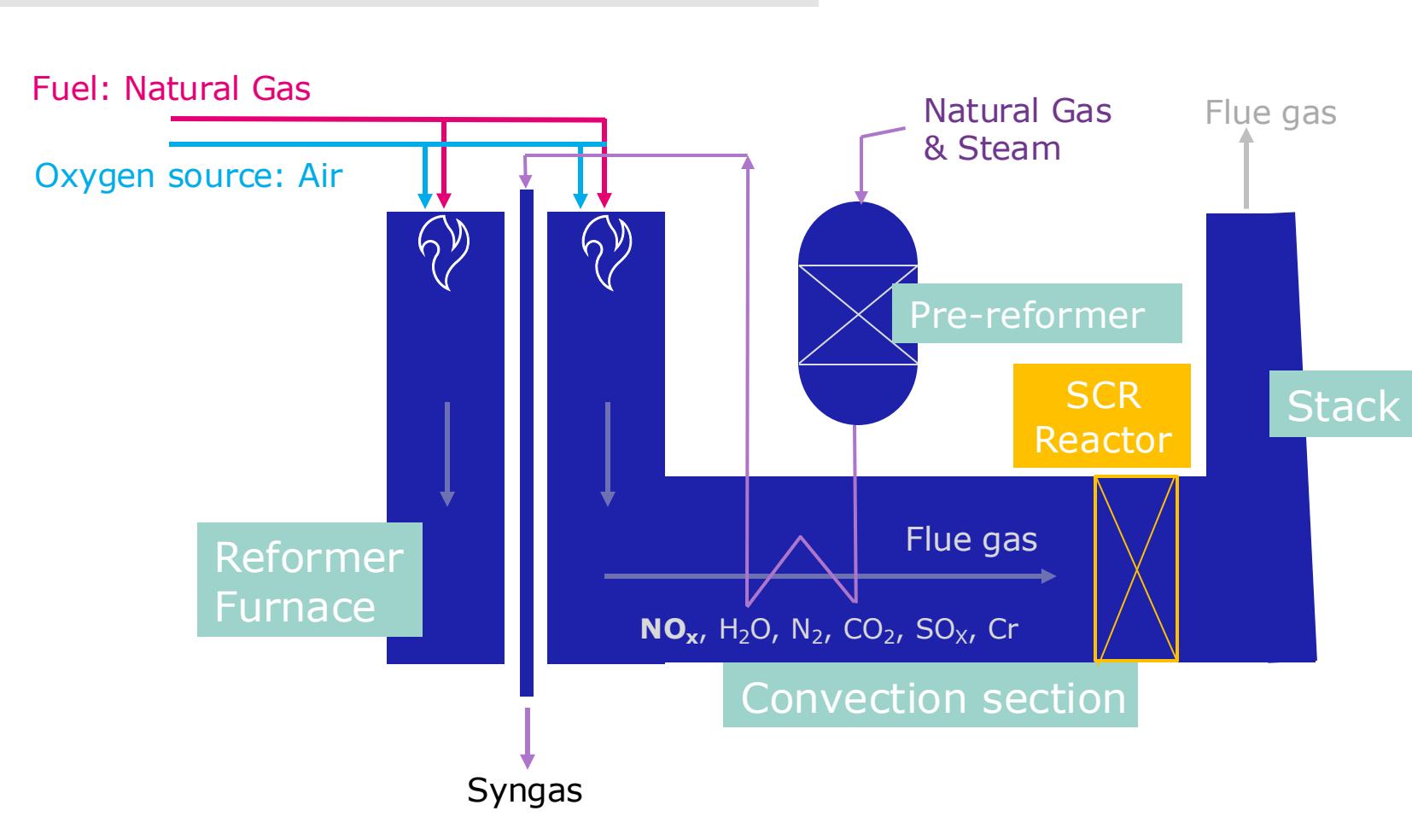
# NO<sub>x</sub> emissions from SMR processes

SMR is an endothermic process.

Heat is required for the SMR reaction to proceed.

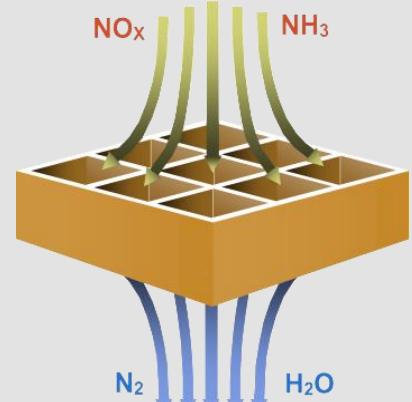
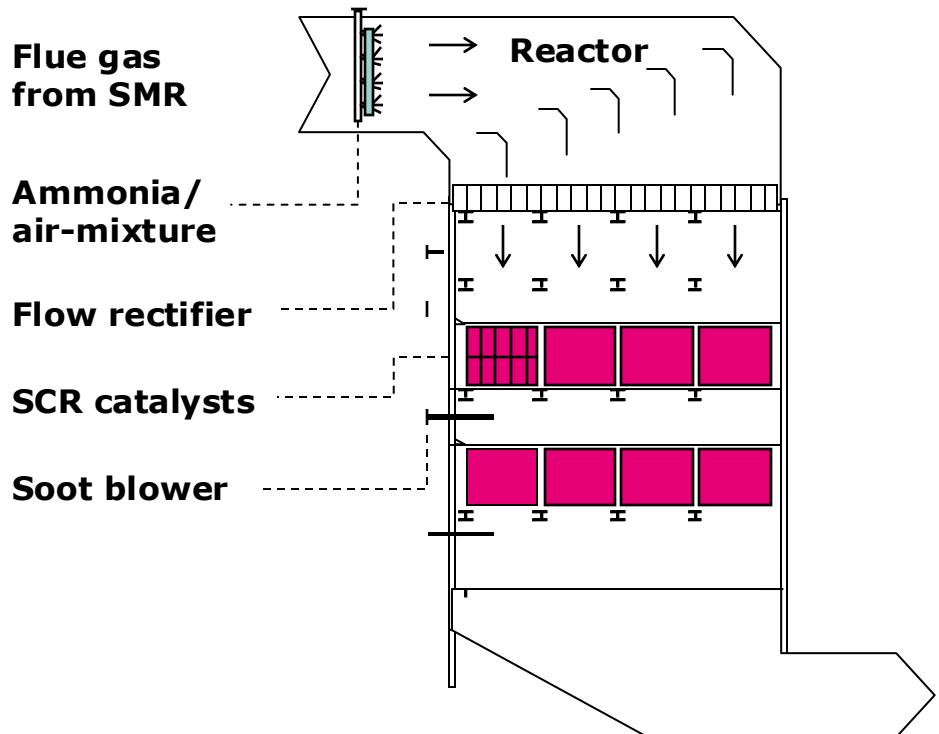
**NO<sub>x</sub> emissions** are formed during the combustion of fuel in the reformer furnace.

Where NO<sub>x</sub> regulations apply, SMR units are typically equipped with **Selective Catalytic Reduction** (SCR) systems.



# Selective Catalytic Reduction (SCR)

**Inlet:**  $\text{NO}_x$ ,  $\text{NH}_3$ ,  $\text{O}_2$ ,  $\text{SO}_x$ ,  $\text{H}_2\text{O}$ ,  $\text{N}_2$ ,  $\text{CO}_2$



**$\text{NO}_x$  is reduced by ammonia across the SCR catalyst:**

- $4\text{NO} + 4\text{NH}_3 + \text{O}_2 \rightarrow 4\text{N}_2 + 6\text{H}_2\text{O}$  (standard)
- $\text{NO} + \text{NO}_2 + 2\text{NH}_3 \rightarrow 2\text{N}_2 + 3\text{H}_2\text{O}$  (fast)
- $2\text{NO}_2 + 4\text{NH}_3 + \text{O}_2 \rightarrow 3\text{N}_2 + 6\text{H}_2\text{O}$  (slow)

**Undesirable side reactions:**

- $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$  (non-selective oxidation)
- $4\text{NH}_3 + 4\text{O}_2 \rightarrow 2\text{N}_2\text{O} + 6\text{H}_2\text{O}$  (non-selective oxidation)
- $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$
- $2\text{NH}_3 + \text{SO}_3 + \text{H}_2\text{O} \rightarrow (\text{NH}_4)_2\text{SO}_4$

# SINOx extruded ceramic honeycomb SCR catalyst

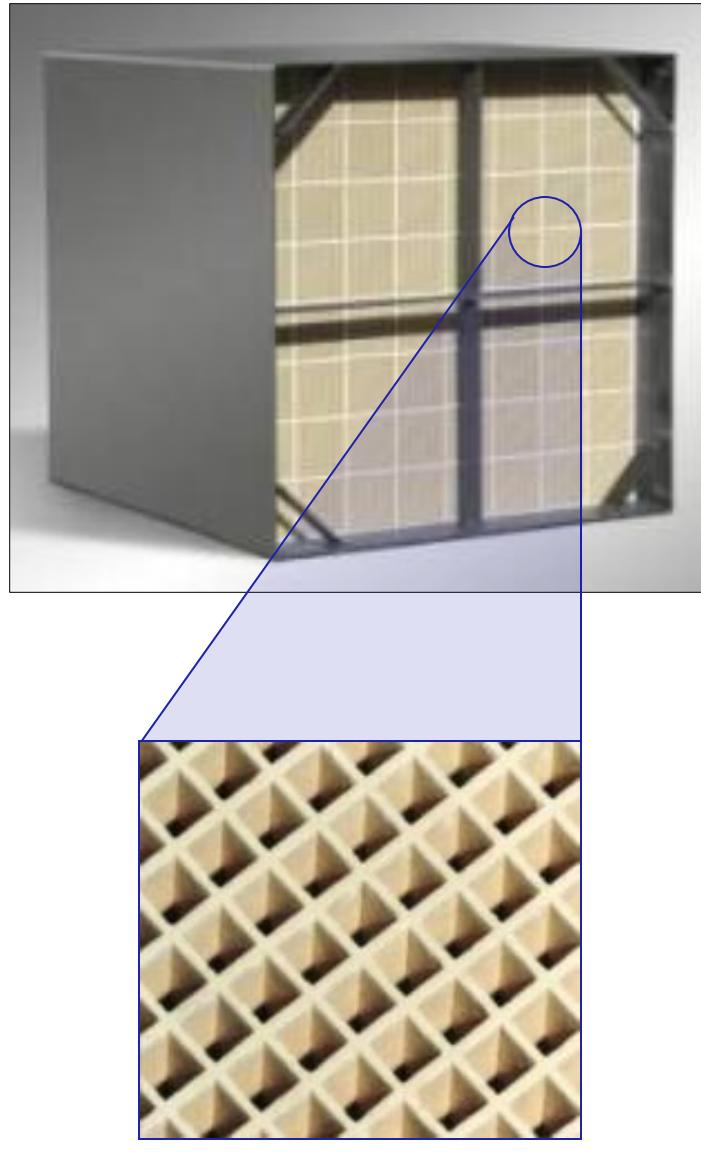
Made entirely of active materials

Low to medium-dust application

High specific surface area

High activity

Variable length and number of cells (6-300 CPSI)



# Relative SCR catalyst operating temperatures

**Higher vanadium loading increases  $\text{NO}_x$  conversion at low temperature**

**Lower vanadium loading is better at higher temperature**

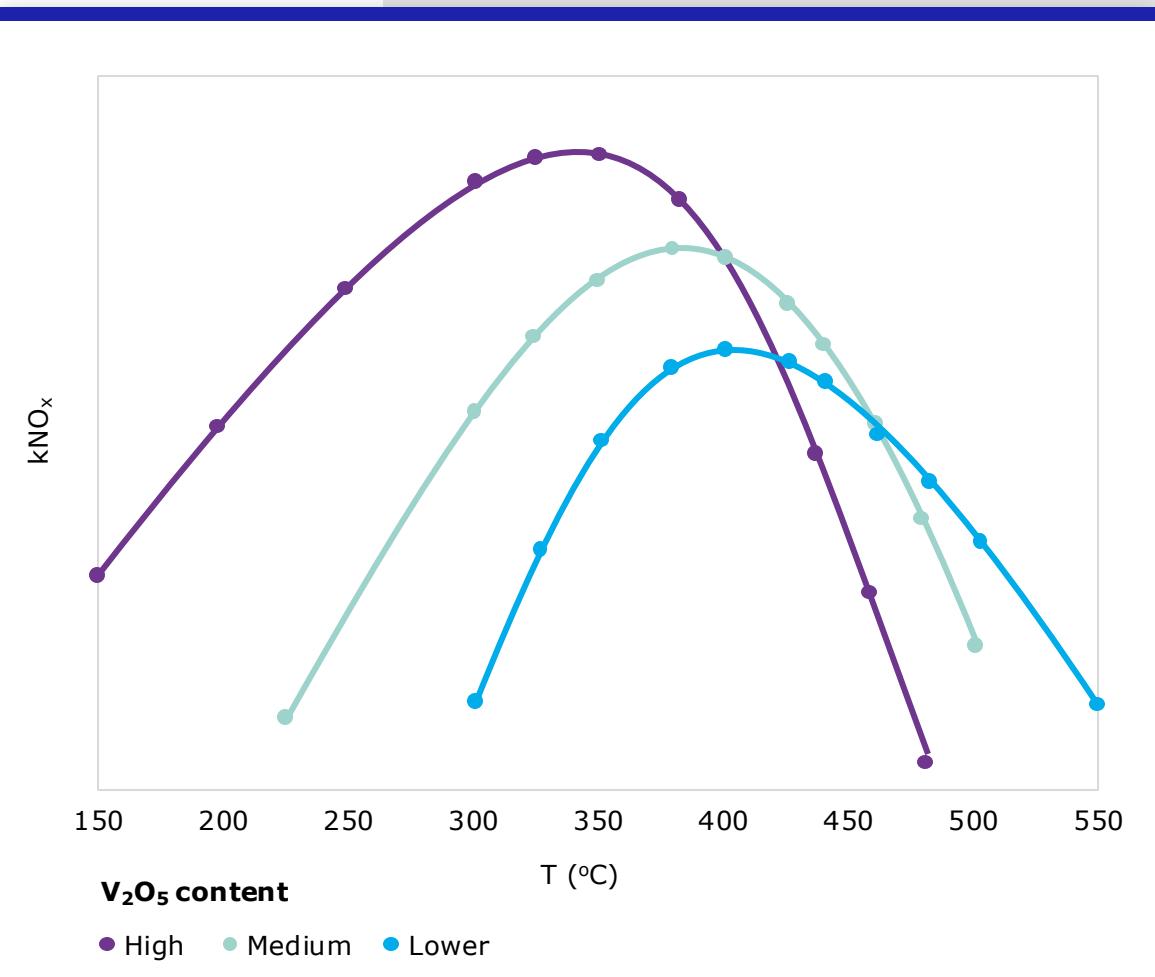
Higher V promotes the undesired reaction:



$$k\text{NO}_x = - \frac{\text{exhaust flow rate}}{(\text{cat volume}) * (\text{GSA})} * \ln(1-x)$$

$x$  = fraction  $\text{NO}_x$  conversion

**GSA** = geometric surface area (function of cell density)



# Typical conditions:

## SMR vs other SCR applications

### Impact on design:

T, O<sub>2</sub> and H<sub>2</sub>O concentrations

NO<sub>x</sub> concentration

Conversion targets

NH<sub>3</sub> Slip

Operating period

Future T window – higher  
>500°C (932°F)

Application	SMR	Gas turbines	Carbon black
Temp. window (°C)	275-450	CC: 300-400 SC: 470-550	300-360
NO <sub>x</sub> conc. (ppm, @ref. O <sub>2</sub> )	30-60	~10-80	200-400
Act O <sub>2</sub> (%)	<b>1.2-5.6</b>	12-16	2.6-5
Ref. O <sub>2</sub> (%)*	3	15	7
H <sub>2</sub> O conc. (%)	<b>16.5-27.8</b>	10	~40
NO <sub>x</sub> conv. target (%)	70-95	80-99	80-94
NH <sub>3</sub> slip (ppm)	1-10	2-10	<5
Guarantee period (oph)	<b>26,000-44,000</b>	8,000/16,000	24,000
Pressure drop limit (mbar)	2.5-10	2	10

\* Depends on local requirements/legislation

**SMR**  
Steam Methane Reforming

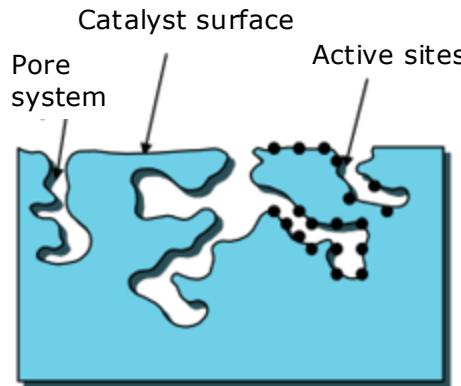
**CC**  
Combined Cycle

**SC**  
Simple Cycle

# Deactivation mechanisms

## Poisoning

Deactivation of the active sites by chemical attack (e.g. alkalis, phosphorus)

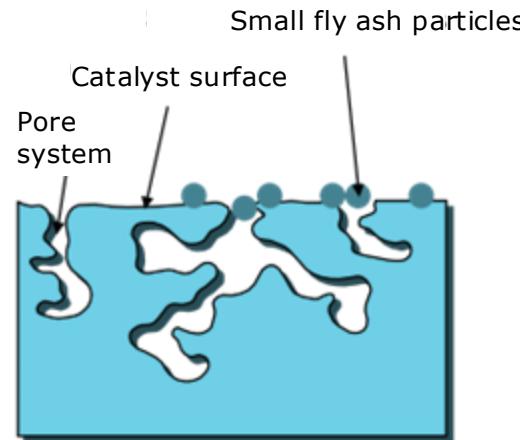


Adsorption of ammonia **inhibited**

DeNO<sub>x</sub>-reactions **aren't possible any more**

## Plugging

Microscopic blockage of the pore system by small fly ash particles

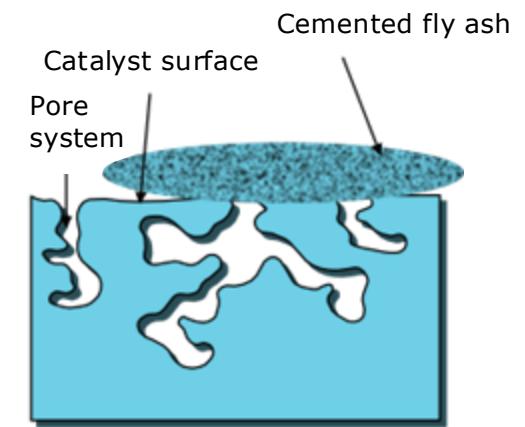


Fly ash particles are **diffusing** into the **catalyst pores**

Catalyst micro pore systems **plugged mechanically**

## Masking

Macroscopic blockage of catalyst surface by cement fly ash



Reactive particles **grow on the surface**

Due to **high amount of calcium oxide in the ash**

# Steam methane reformer field experience

**Main driver for catalyst deactivation in Steam Methane Reformer (SMR)**

Masking

Poisoning



**Fresh - JM** → **40K hrs SMR - JM**





**Analytical methods:**

**Chemical analysis** – surface vs. bulk composition  
**Physical parameters** – surface area, pore volume  
**Activity testing** (fresh vs. aged)



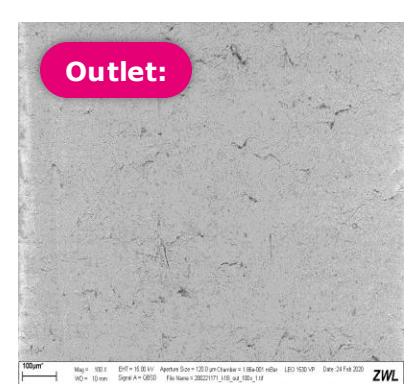
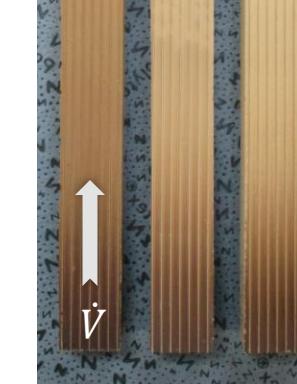
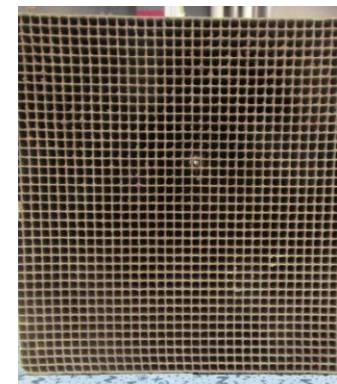
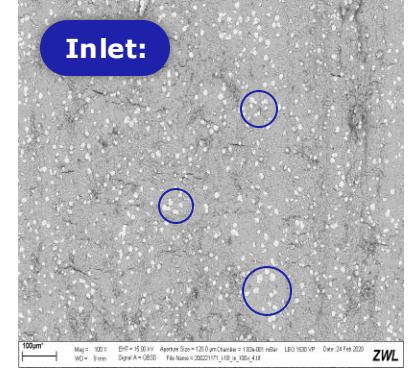
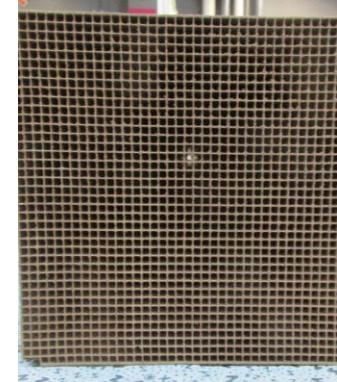
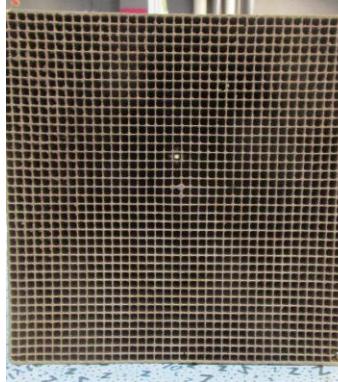
**JM**

The data included herein were collected in a Johnson Matthey laboratory which has not been certified by the relevant authorities/agencies to perform emissions testing. These are indicative data and do not represent a guarantee that the tested catalyst or emissions system will pass the relevant emissions legislation.

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# Visual discolouration of field returns due to poisons and ash

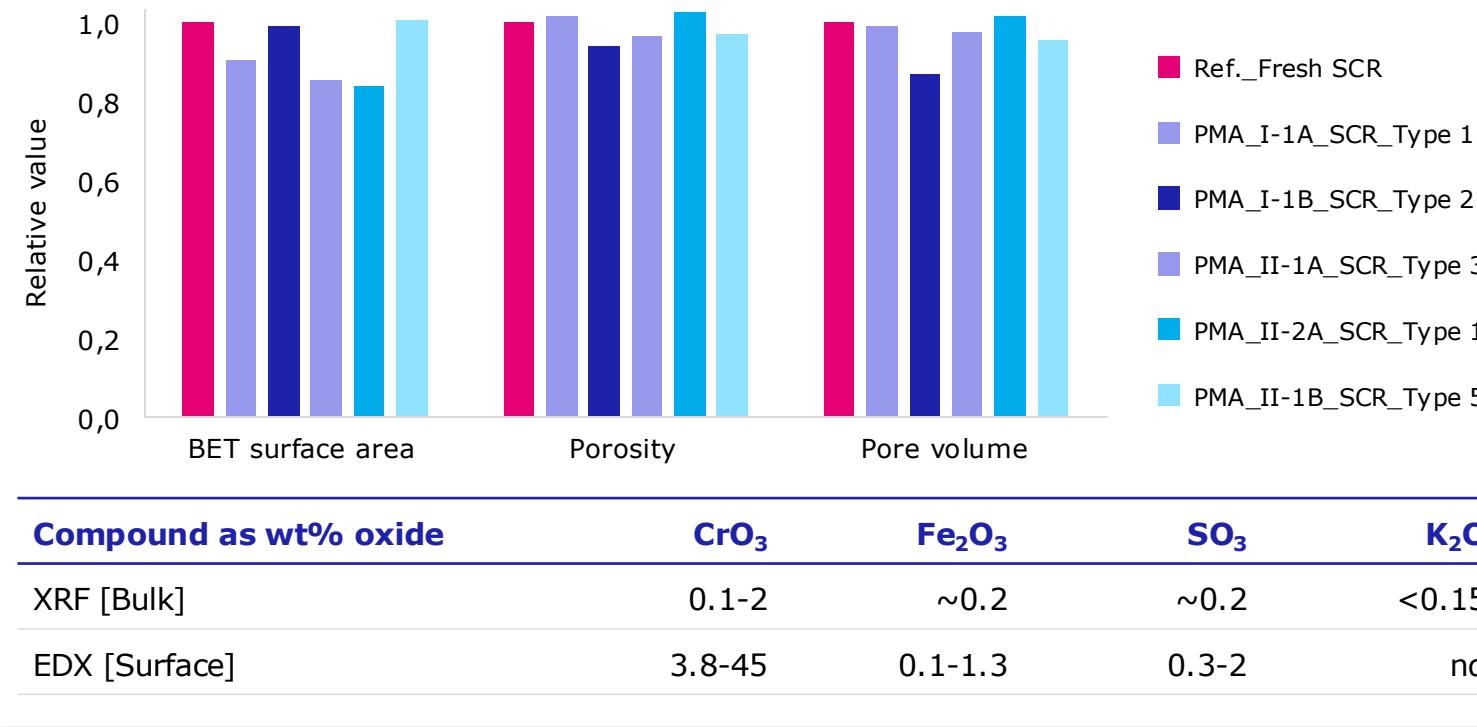


**Discolouring** on the shell, face side and inner channels (inlet side) was shown

Deposition of **particulate ash** on inner walls of inlet side found

**No particulate ash deposition** at outlet side of the catalysts observed

# Combination of physical-chemical techniques used to identify deactivation mechanisms



**No signs of thermal aging or pore plugging**

BET, PR distribution provides insight into thermal aging and/or pore plugging

**XRF: bulk chemical analysis**

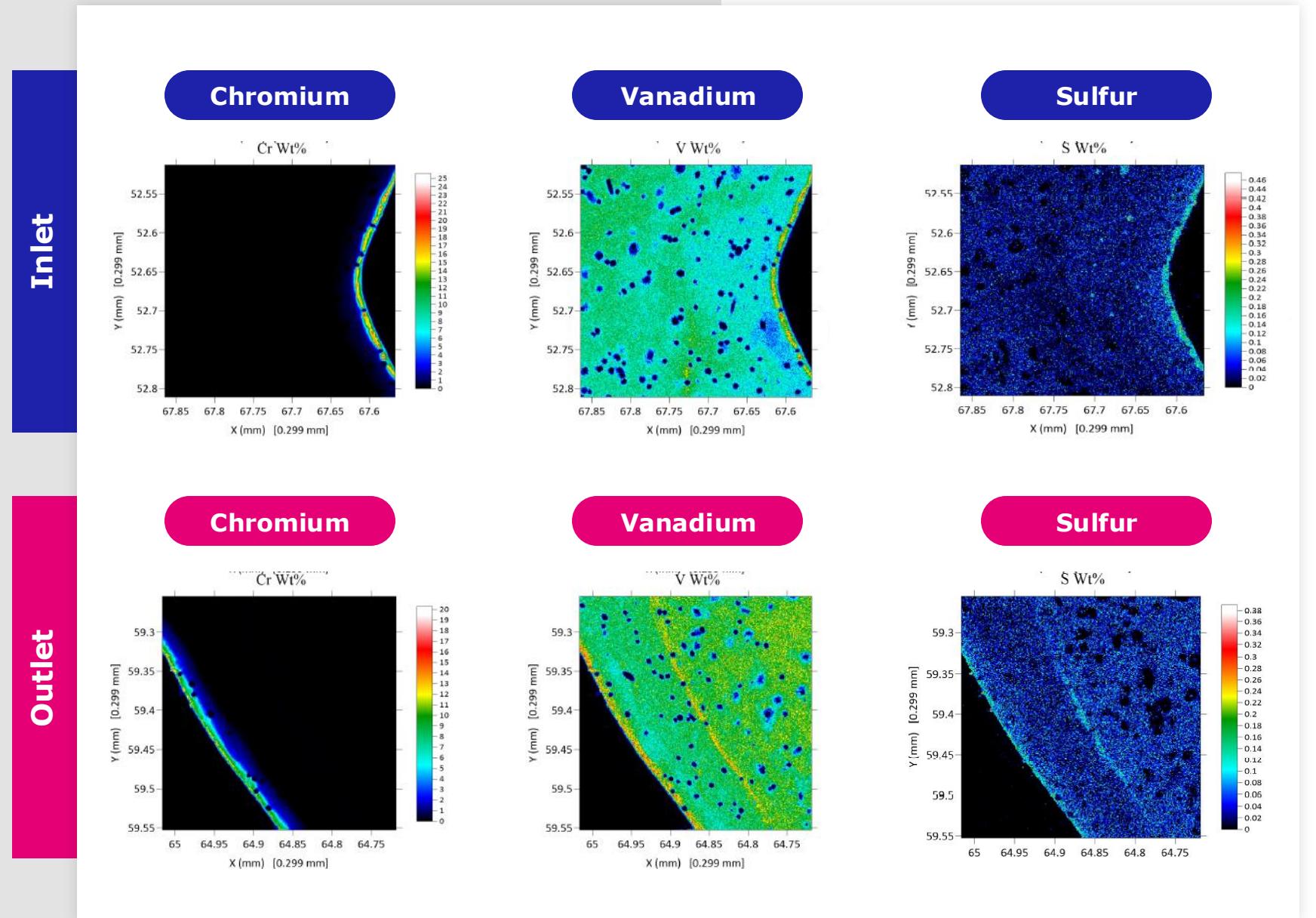
**EDX: surface chemical analysis**

► Where are deposits/poisons located

► Several mechanisms may overlap

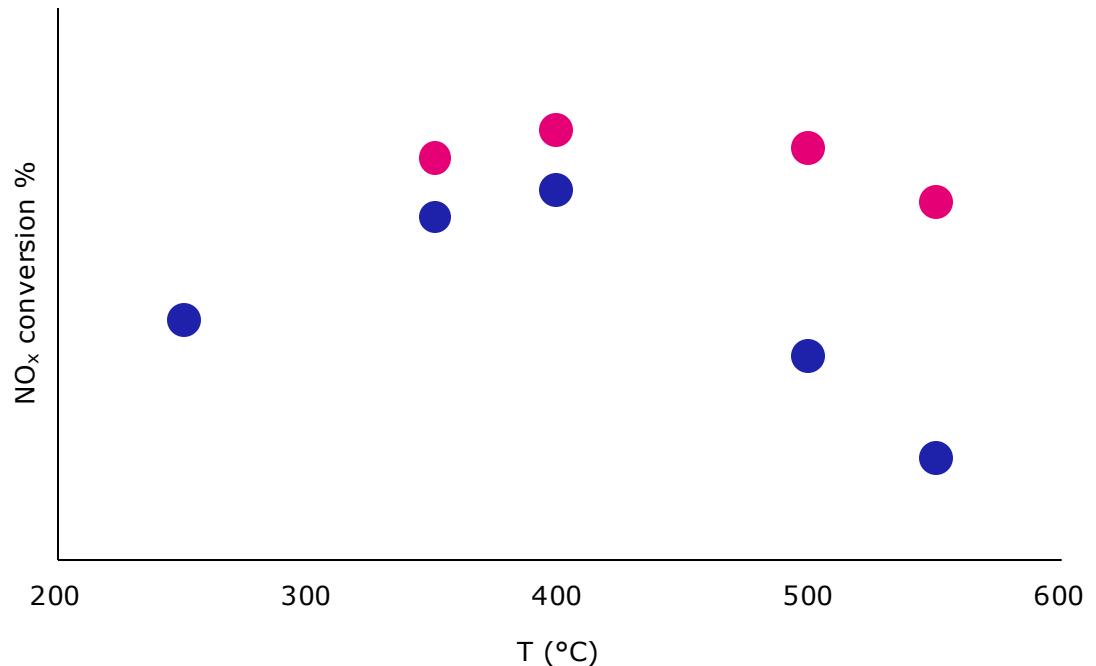
⇒ Cr the primary driver

# Elemental mapping by EPMA reveals Cr concentrates at surface

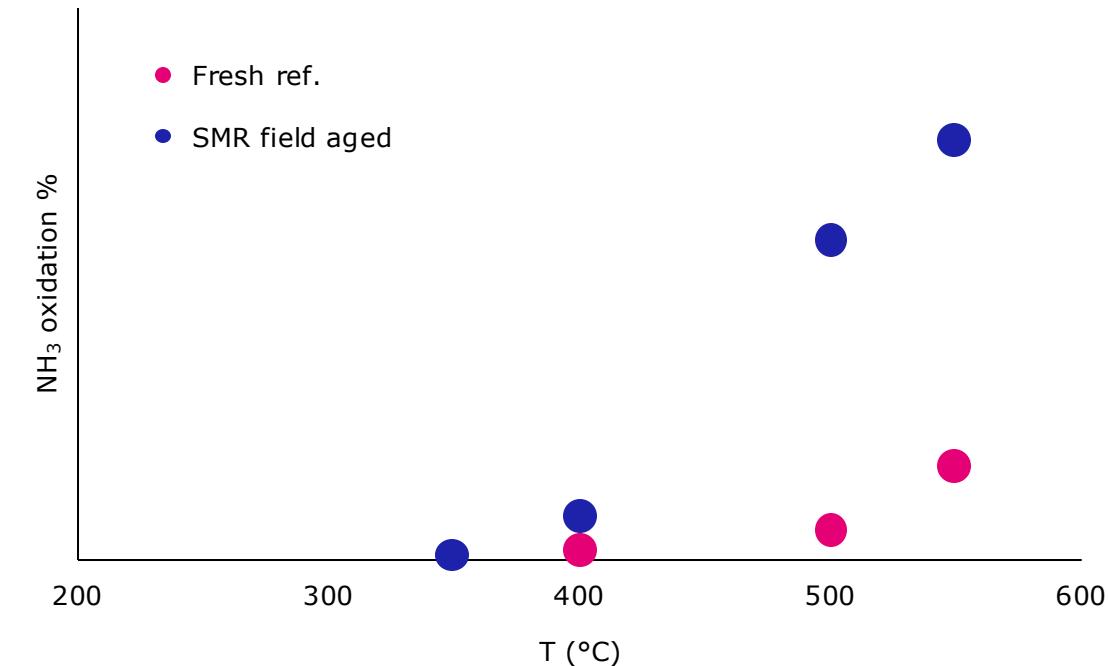


# Catalytic activity negatively influenced by Cr deposition

## Impact of Cr – $\text{NO}_x$ conversion

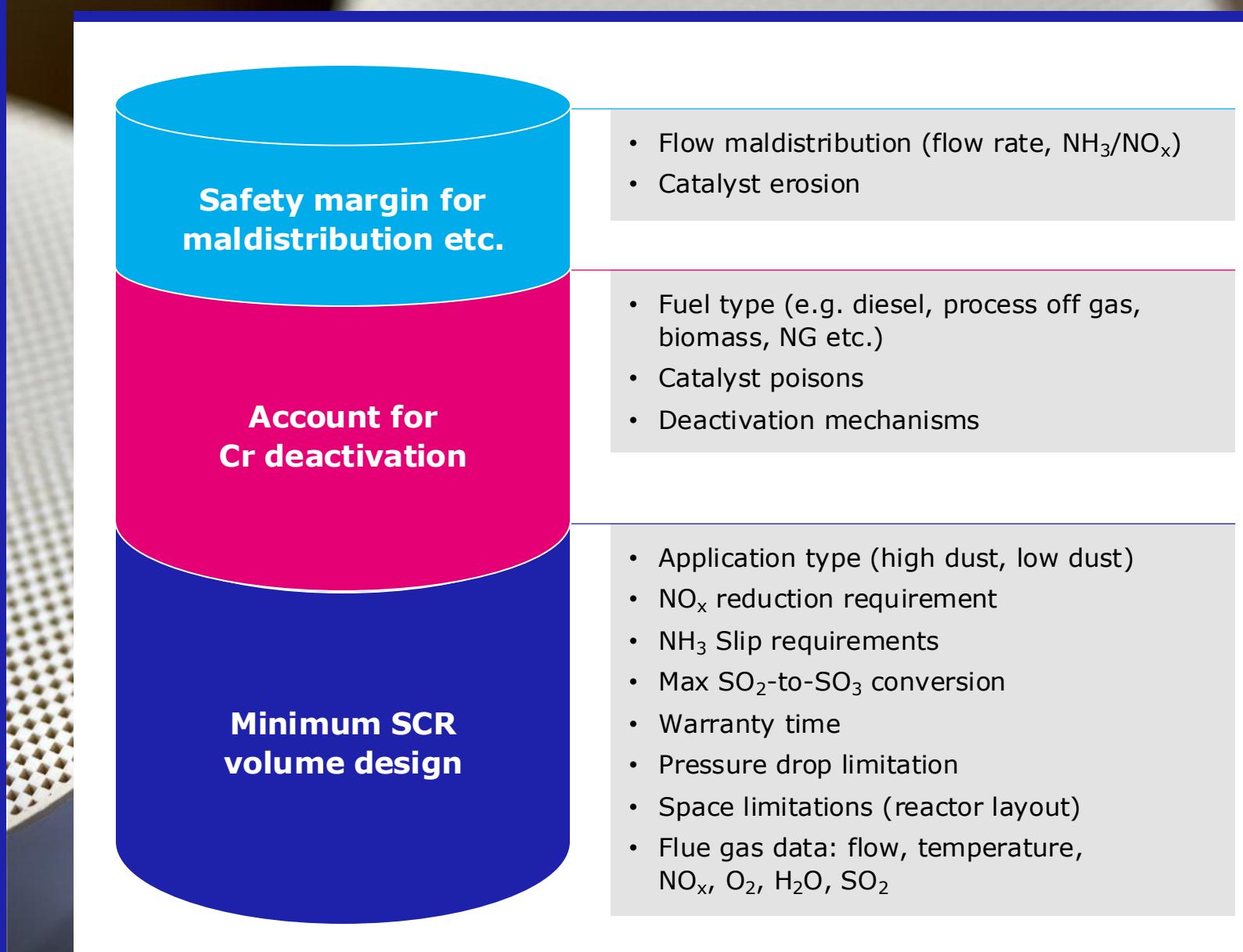


## Impact of Cr – $\text{NH}_3$ oxidation



**Chromium (Cr) – the higher the T, the higher the impact**

# SCR catalyst design considerations for SMR applications



# Innovative solutions to enhance NO<sub>x</sub> performance

## High temperature SCR catalyst

**Increased** system efficiency

**Higher** application temperatures

**Reduced** CO<sub>2</sub> footprint

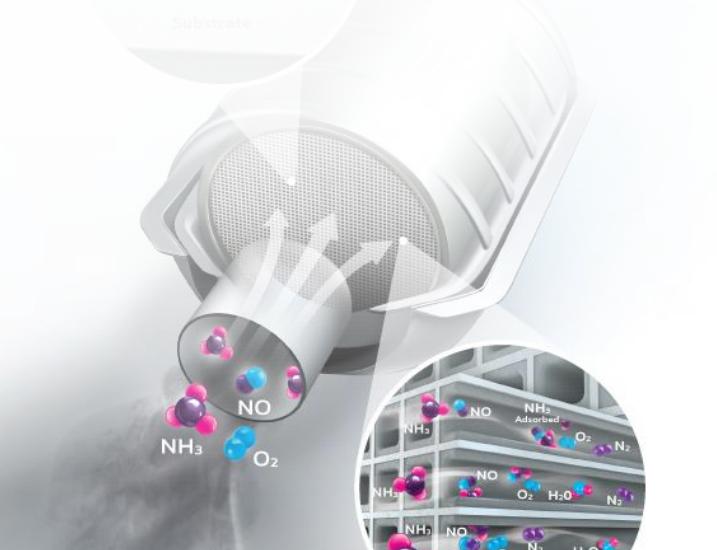


## Ammonia slip catalyst

**Low** NH<sub>3</sub> slip, high NO<sub>x</sub> conversion

Allows for **higher ANR** (overdosing)

**Helps** with flow maldistribution



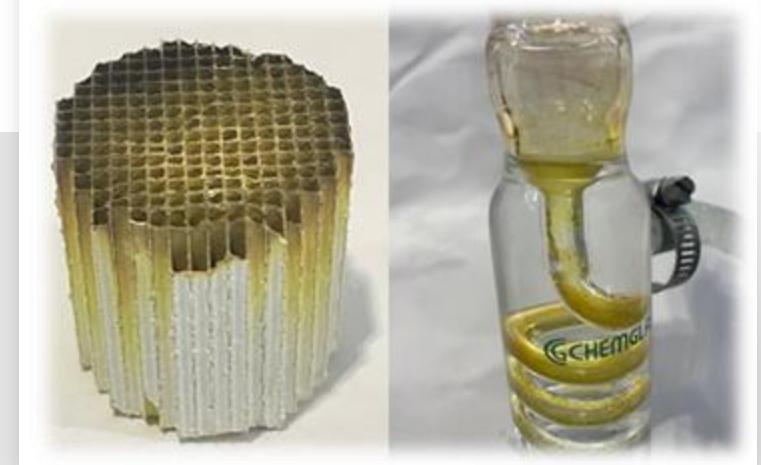
## Chromium-getter

**Protects** downstream catalysts

**Improves** plant performance

**Reduces** early shutdowns

We can provide a **solution!**



# SCR High Temperature catalyst (SINOx-HT):

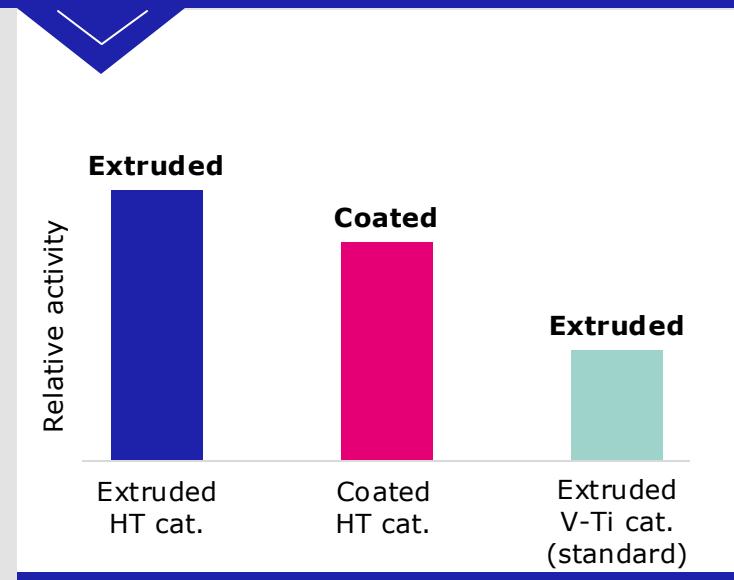
Developed for operation between 450°C-650°C (842°F-1202°F)



# Advanced high temperature SCR catalyst (SCN7000) significantly outperforms standard V-Ti catalyst and washcoated catalysts

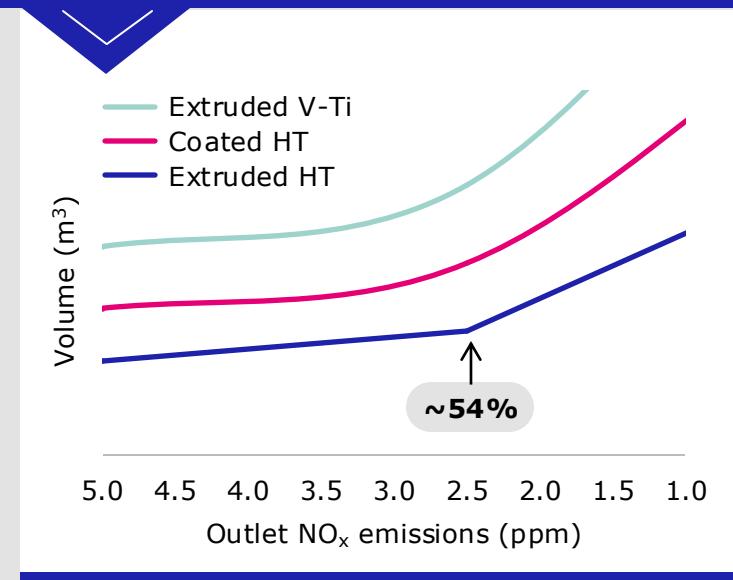
## SCR activity

T=550°C (1022°F); NH<sub>3</sub>/NO<sub>x</sub>=1



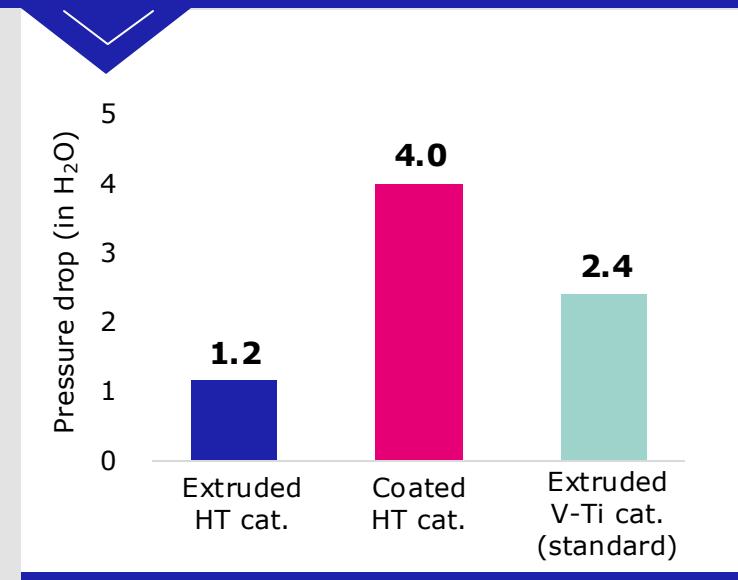
## Catalyst volume

T=550°C (1022°F); NH<sub>3</sub>/NO<sub>x</sub>=1



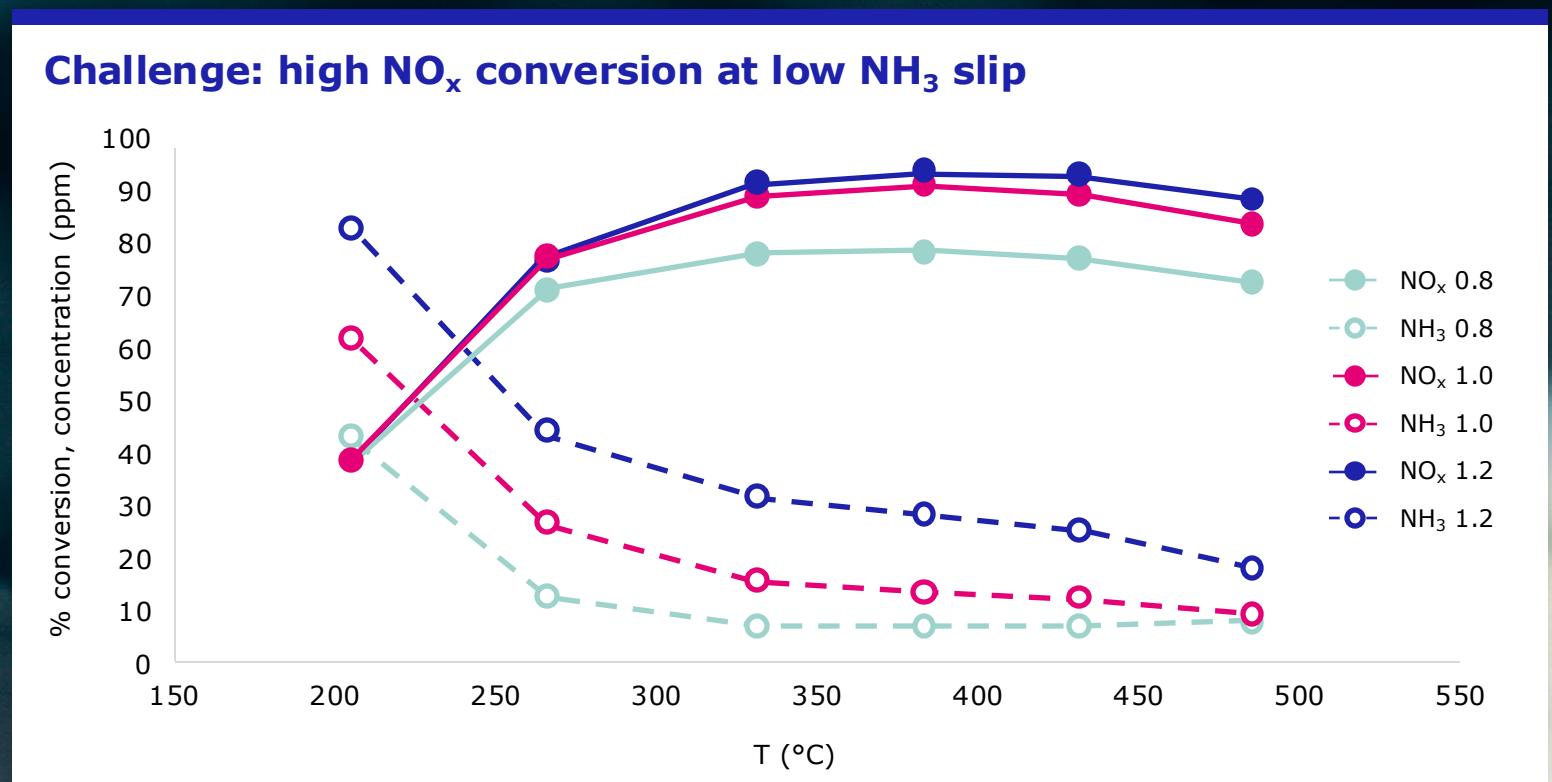
## Pressure drop

T=550°C (1022°F); NH<sub>3</sub>/NO<sub>x</sub>=1



**Superior SCR activity, reduced catalyst volume, and lower pressure drop aligned to tightened emission regulations of next generation power systems**

Ammonia Slip Catalyst (ASC) can boost  $\text{NO}_x$  conversion, reduce  $\text{NH}_3$  slip, compensate for non-uniform  $\text{NH}_3$  distribution, and oxidize CO/VOCs



**Non-uniform  $\text{NH}_3$  distribution can result in localized ANRs**

ANR < 1 results in incomplete  $\text{NO}_x$  conversion

ANR > 1 results in  $\text{NH}_3$  slip

**Non-uniform  $\text{NH}_3$  distribution can be a result of:**

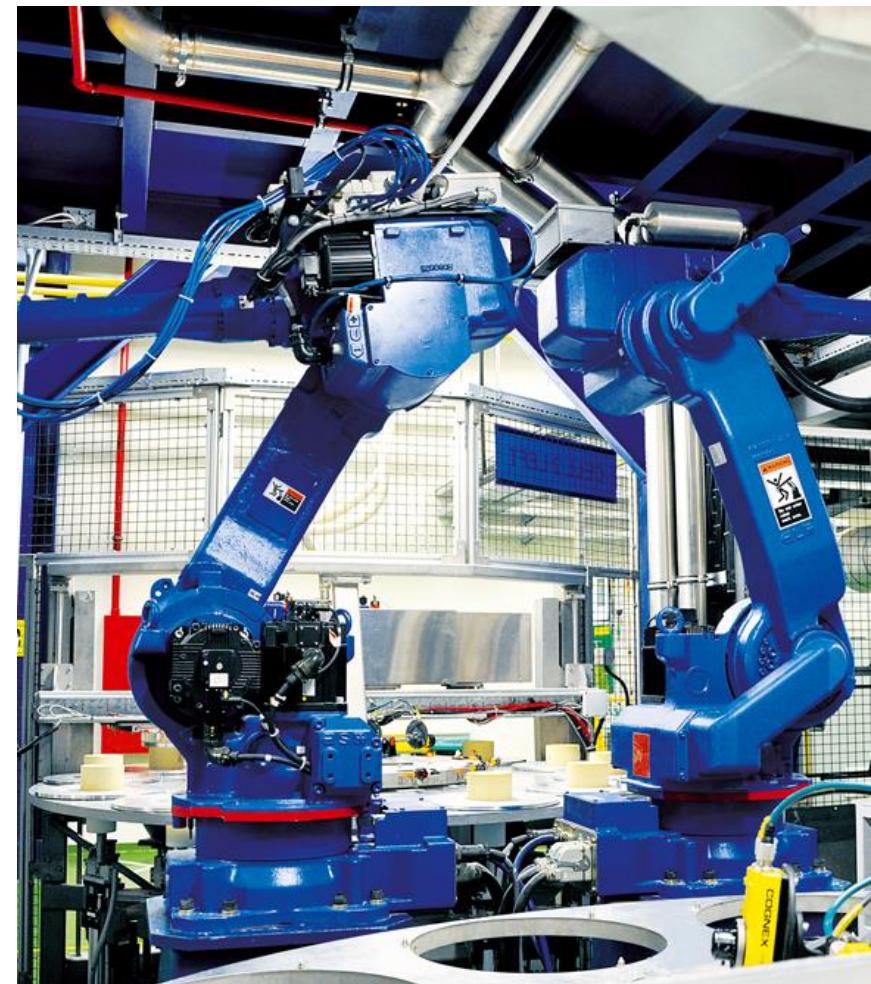
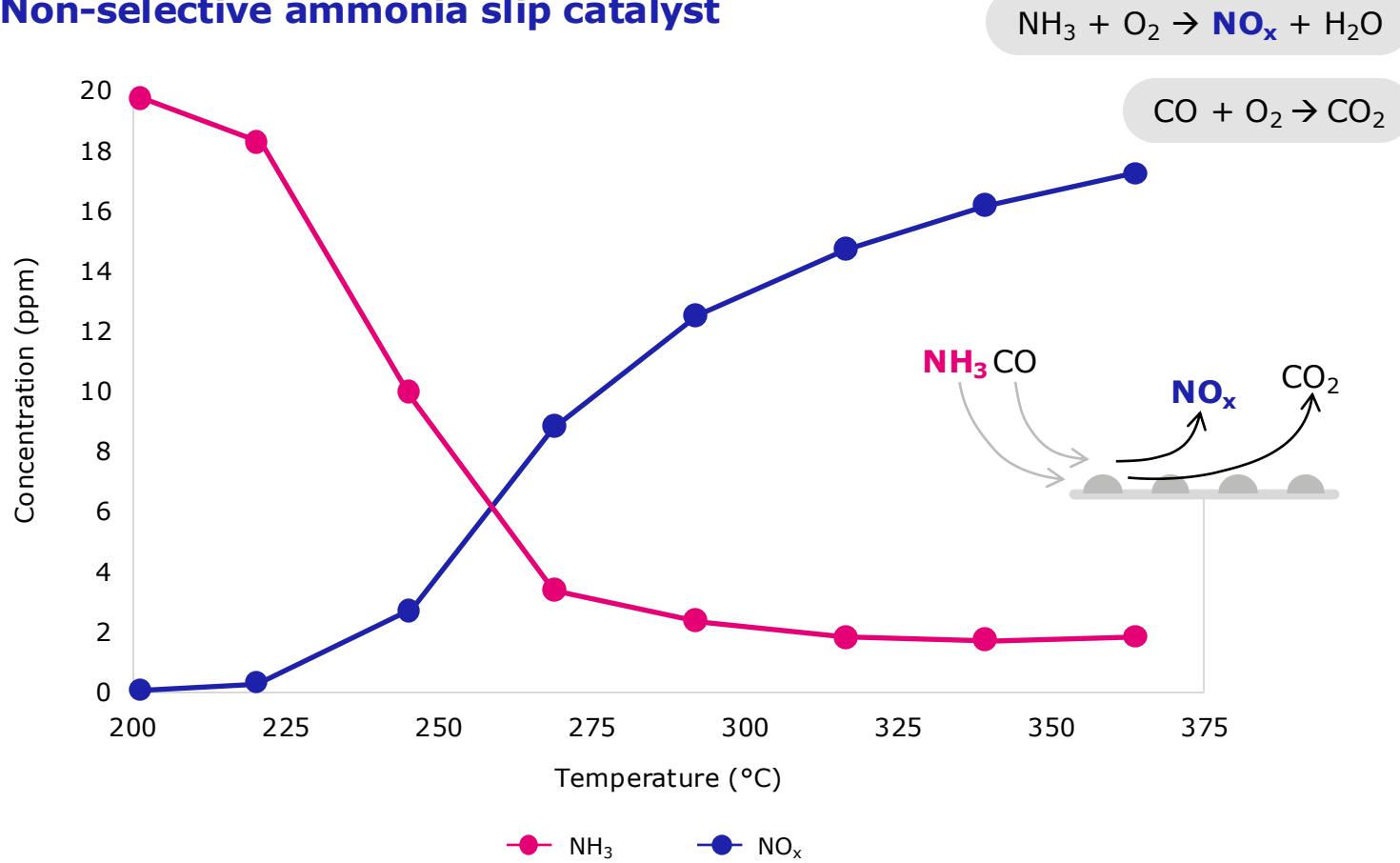
Control system

Gas flow characteristics

Fluctuating load

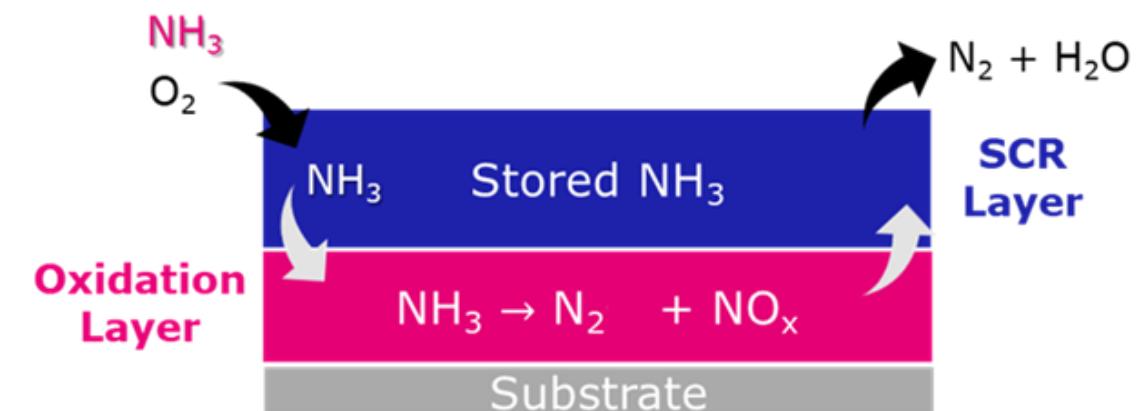
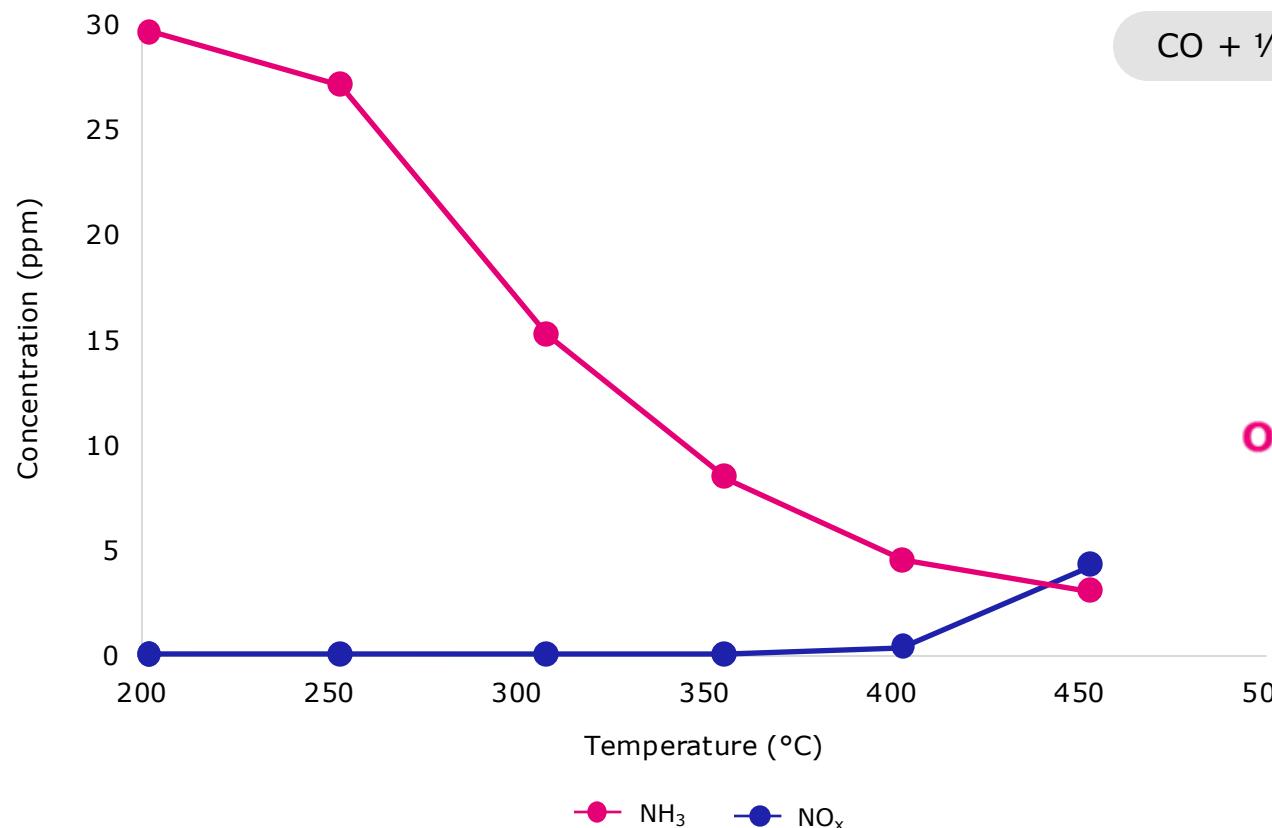
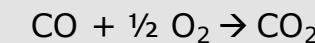
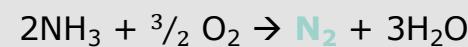
# Previous generation ASC exhibit excellent activity (High NH<sub>3</sub>/CO conversion) but poor selectivity (NO<sub>x</sub> production)

## Non-selective ammonia slip catalyst



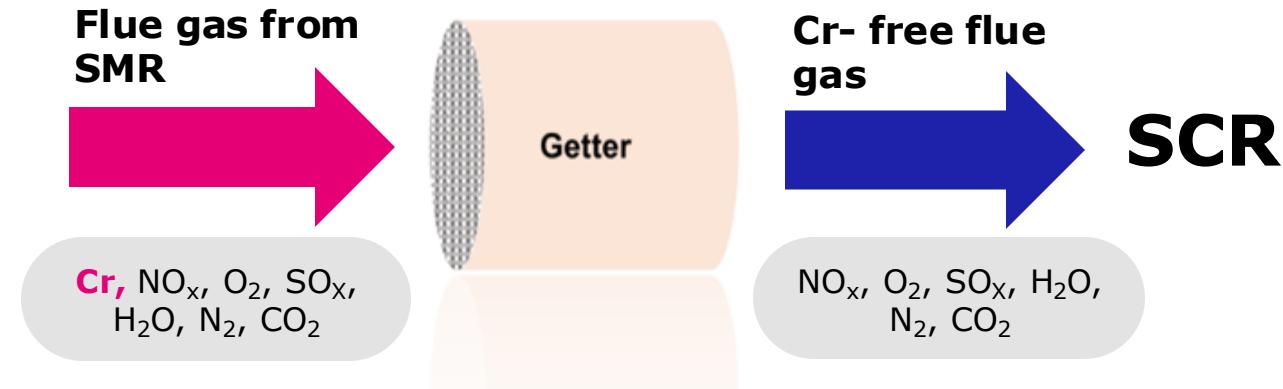
# Advanced ASC performs both oxidation function and SCR function (selective to N<sub>2</sub>) simultaneously

## Selective ammonia slip catalyst



# Chromium Getter

Identified solution to trap Cr vapour released from steel at high temperatures



## Chromium Poisoning

Cr can deposit on downstream catalysts (SCR, ASC), killing performance efficiency

Hexavalent chromium is also carcinogenic

## JM's solution

Cr getters (coated ceramic/metallic monoliths) can be positioned upstream of the SCR

94-99% Cr-capture efficiency proven at lab scale.

# Durable Chromium Getter Technology Developed and Proven Under High Cr Flux Conditions (Condensate)

## No Cr-Getter



0 hrs



100 hrs



300 hrs

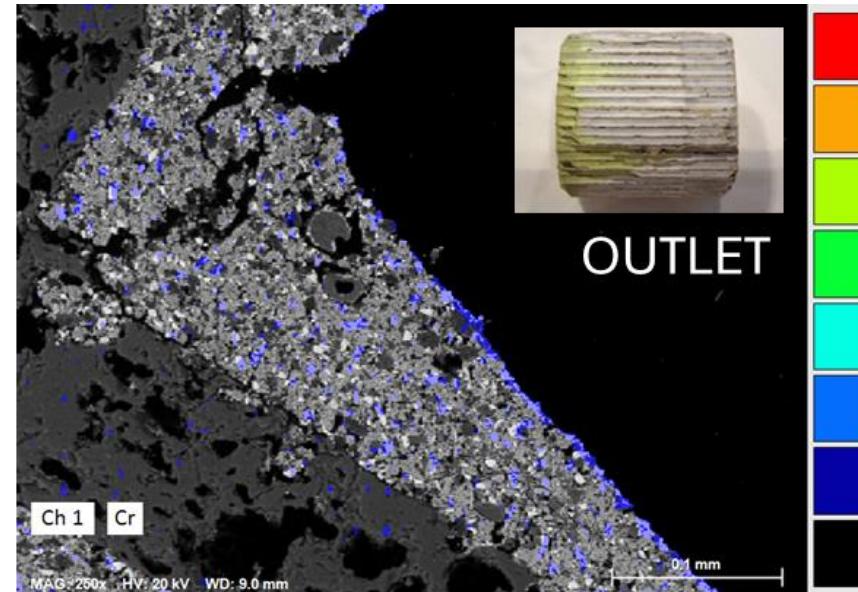
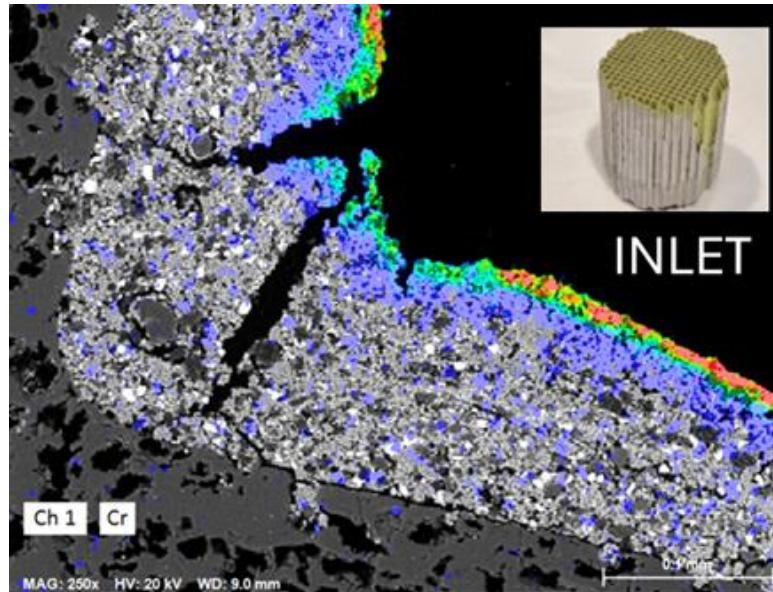
Testing Time  
High Cr Flux



## With Cr-Getter



# Durable Chromium Getter Technology Developed and Proven Under High Cr Flux Conditions (Evidence from SEM/EDX)



**Cr Trapping at Inlet Face (~5mm) as Designed = Longer Lifetime**

**Upstream Cr Trap should improve performance and extend catalyst lifetime**

Cr Getter must be upstream of NH<sub>3</sub> injection as Cr can act as oxidation catalyst

High capture efficiency and suitable sizing design should offer protection over catalyst lifetime

Improved SCR and ASC function from Cr protection should reduce aggregate emissions

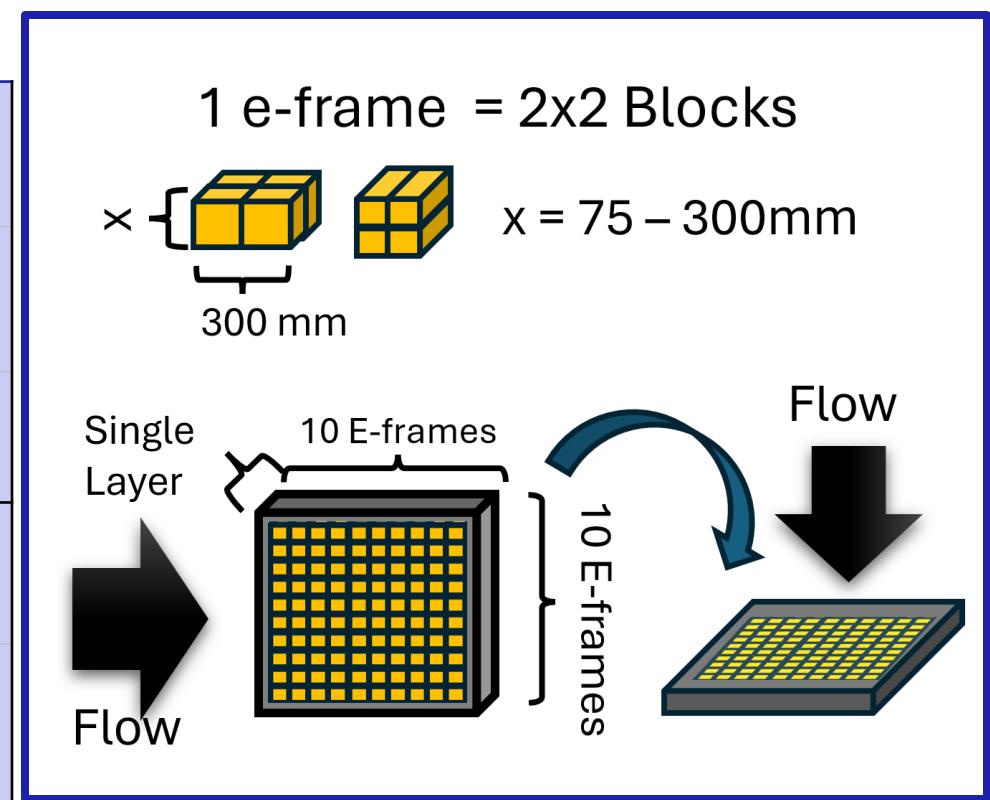
# Cr Getter Sizing Considerations

Key Parameters Dictating Size		Parameter	Sizing Example 1	Sizing Example 2	Sizing Example 3
<b>Operating Temperature (550-750+ °C recommended)</b>		Mass Flow Rate (kg/hr)	10,000 kg/hr	75,000	75,000
<b>Cr Slip Requirement</b>		Temperature (°C)	550 °C	550 °C	750 °C
<b>Available Cross Section &amp; Total Volume</b>		Reactor Cross Section (m <sup>2</sup> )	10 m <sup>2</sup>	45 m <sup>2</sup>	45 m <sup>2</sup>
<b>Guarantee period</b>		Total Volume (m <sup>3</sup> )	1.4 m <sup>3</sup>	6.3 m <sup>3</sup>	6.3 m <sup>3</sup>
<b>Pressure Drop Requirement</b>		Predicted Pressure Drop (in H <sub>2</sub> O)	< 0.3 in H <sub>2</sub> O	< 0.4 in H <sub>2</sub> O	< 0.7 in H <sub>2</sub> O

# Cr Getter Sizing Considerations

Key Parameters Dictating Size	
<b>Operating Temperature</b> (550-750+ °C recommended)	
<b>Cr Slip Requirement</b>	
<b>Available Cross Section &amp; Total Volume</b>	
<b>Guarantee period</b>	
<b>Pressure Drop Requirement</b>	

Parameter	Sizing Example 1
Mass Flow Rate (kg/hr)	10,000 kg/hr
Temperature (°C)	550 °C
Reactor Cross Section (m <sup>2</sup> )	10 m <sup>2</sup>
Total Volume (m <sup>3</sup> )	1.4 m <sup>3</sup>
Predicted Pressure Drop (in H <sub>2</sub> O)	< 0.3 in H <sub>2</sub> O



# Summary

**Selective catalytic reduction (SCR) catalysts used to reduce NO<sub>x</sub> emissions**

- Formulation defined based on SCR design temperature
- Next generation SCR technology for high temperatures (450°C-650°C) commercially available

**Ammonia slip catalyst (ASC) allows continuous operation at higher NH<sub>3</sub>/NO<sub>x</sub> ratios (ANRs)**

- Results in higher NO<sub>x</sub> conversion while maintaining low NH<sub>3</sub> slip
- Improve plant performance by reducing back-end deposits (saves O&M costs for removal)
- Active for CO and/or VOC oxidation
- Can help compensate for non-ideal NH<sub>3</sub> distribution

**Chromium (Cr) deposition is primary failure mode for steam methane reformer (SMR) applications**

- Cr deposits primarily on the catalyst surface
- Results in increased ammonia (NH<sub>3</sub>) oxidation
- Careful design considerations required to meet lifetime requirements
- Cr-getters being developed to protect downstream catalysts

JM Johnson  
Matthey